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3,196,129
INYL POLYMERS STABILIZED WITH THIO-ACETAL AND THIOKETAL TIN COMPOUNDS Ingenuin Hechenbleikner, Kenwood, Robert E. Bresser, Sharonville, and Otto A. Homberg, Woodlawn, Ohio, assignors to Carlisle Chemical Works, Inc., Rending, Ohio, a corporation of Ohio

No Drawing. Filed May 1, 1962, Ser. No. 191,463 18 Claims. (Cl. 260-45.75)

This application is a continuation-in-part of application Sorial No. 103,256, filed April 17, 1961, now Patent : 1378.290, issued February 19, 1963.

The present invention relates to novel thioacetals and thioketals and their use as stabilizers for solid polymers 15 monoolefins having 2 to 4 carbon atoms and as stailizers for solid polymers of monoolefins having 2 to 4 urbon atoms and as stabilizers for halogen-containing uns.

It is an object of the present invention to prepare novel 20 ilioacetals and thioketals.

Another object is to prepare novel mono-, di- or tri-Jrocarbon tin salts of acids having a thicketal or thicetal group.

An additional object is to prepare novel stabilized com- 25 titions containing solid polymers of monoolefins having ... 4 carbon atoms, preferably polypropylene.

Another object is to stabilize polypropylene and other lymers of monoolefins having 2 to 4 carbon atoms with ergistic stabilizer compositions.

further object is to extend the heat and light stability logen-containing resins.

an additional object is to provide novel stabilized vinyl in compositions.

Yet another object is to provide novel synergistic sta- 35 we zers for halogen-containing resins.

Still further objects and the entire scope of applicability on the present invention will become apparent from the detailed description given hereinafter; it should be understood, however, that the detailed description and specific 40 vicine n, R, R₂ and R₃ are as defined above. examples, while indicating preferred embodiments of the invention, are given by way of illustration only, since various changes and modifications within the spirit and scope of the invention will become apparent to those skilled in the art from this detailed description.

In one form of the invention there are prepared dihydrocarbon tin salts of acids having a thioacetal or thioketal grouping. Such compounds belong to one of the following groups

 G_{2}

where n is an integer from 1 to 8, R and R₁ are alkyl, resulkyl or aryl and R2 and R3 are hydrogen, alkyl, alkenyl, i, aryl, hydroxyaryl, alkoxyaryl or taken together etc a cyclohexane ring, i.e., the pentamethylene JCi.

where n, R and R₁ are as defined above, R₄ and R₅ are allayl, aralkyl, aryl, carboxylic acid or carboxylic acid wer and Re is hydrogen, alkyl, benzyl or aryl.

where R, R, R, and R, are as defined above

where n, R and R₁ are as defined above and R₇ is alkyl, acalkyl or aryl.

where R, R₁, and n are defined as above, R₂ is alkyl, :r: kyl or aryl and m is an integer from 0 to 8.

Another aspect of the invention is the preparation of menohydrocarbon and trihydrocarbon tin salts of acids having a thioacetal or thioketal grouping. Such comprounds belong to one of the following groups:

re n, R, R4, R5 and R6 are as defined above.

v icre R, R₂ and R₃ are as defined ab ve. **0**5 (1) R8n(OOC(CHa) SCH_SR1)1

55

60

70

V ... re n, R, and R_7 are as defined above.

GOVERNMENT

OT

(VII)

10

15 (LX)

70

- Sn- 5

Preferably, R₁₈ is

where R, R_1 , R_2 , R_3 and R_{10} are as defined above and R_{11} , R_{12} and R_{13} are selected from the green consisting 20 of hydrogen and

where n, R, R_1 , R_4 , R_5 , R_6 and R_{10} are as d fined above. 30

where R, R_1 , R_2 , R_3 and R_{10} are as defined above, and R_{14} is selected from the group consisting of hydrogen and

where n, R, R_1 , R_7 and R_{10} are as defined above.

where m, n, R, R_1 , R_8 and R_{10} are as defined above, and R_{15} , R_{16} and R_{17} are selected from the group consisting of hydrogen and

While the formulae are written for the monomeric materials, many of them also exist in a po's meric form closely approximating the monomeric formulae.

In another form of the invention certain thioacetals and thioketals are mixed with dihydrocarbon the oxides or sulfides or with mon hydrocarb n stannoic ands or m no-hydrocarbon tin alcoholates or trihydrocard at in oxides.

The thioacetals and thioketals have me f llowing

formulae:

(VI)

(VIe) H OCCH.
HO C-CH-S R

In Formulae VI through X the definitions of m, n, R₂ R₂, R₄, R₅, R₆, R₇ and R₈ are the same as those defined above and R₉ is hydrogen or alkyl.

As the dihydrocarbon tin oxide or sulfide, there can be used dimethyl tin oxide, dibutyl tin oxide, dioctyl tin oxide, dilauryl tin oxide, butyl lauryl tin oxide, dioctadecyl tin oxide, diphenyl tin oxide, dimethyl tin sulfide, dibutyl tin sulfide, dioctyl tin sulfide, dilauryl tin sulfide, diphenyl tin sulfide, dioctadecyl tin sulfide, licyclohexyl tin oxide, etc.

As monohydrocarbon stannoic acids and monohydrocarbon tin alcoholates there can be used methyl stannoic acid, butyl stannoic acid, lauryl stannoic acid, octadecyl stannoic acid, phenyl stannoic acid, butyl tin triisopropylate, octyl tin triisopropylate, octadecyl tin trimethylate, 40 butyl tin tributylate.

As trihydrocarbon tin oxides there can be used tributyl tin oxide [(C₄H₉)₃Sn]₂O, trioctyl tin oxide, trioctadecyl tin oxide, triphenyl tin oxide, dibutyl octyl tin oxide, and trilauryl tin oxide.

Typical examples of compounds in Group I are dibutyl tin propane-2,2-bis (mercaptopropionate)

dibutyl tin benzaldi (mercaptoacetate),
dibutyl tin benzaldi (mercaptopropionate),
dibutyl tin 2-hydroxybenzaldi (mercaptopropionate),
dineopentyl tin propane-2,2-bis mercaptopropionate),
dibutyl tin isodecane 1,1-bis mercaptopropionate),
dioctyl tin propane 2,2-bis (mercaptopropionate),
dibutyl tin cyclohexyl 1,1-bis (mercaptopropionate),
dimethyl tin propane 1,2-bis (omega mercaptooctanoate),
dioctadecyl tin methane bis (mercaptoacetate),
diphenyl tin propane 2,2-bis (mercaptopropionate),
dibutyl tin propene 3,3-bis (mercaptopropionate),
butyl lauryl tin phenylacetaldi (mercaptoacetate),
dihexyl tin 2-butene-1,1-bis (mercaptopropionate),
dibutyl tin diphenyl methane bis (mercaptopropionate)

dibenzyl tin propane-2-mercaptoacetate-2-mercaptopropionate, dibutyl tin 2-hydr xy-4-methoxybenzaldi (mercaptobutyrate) and 75 dibutyl tin pr pane 2,2-bis (mercaptosuccinate). . Typical examples of compounds in Group II are tyl tin di [4,4-bis (dodecylthio) valerate]

dibutyl tin di [4,4-bis (phenylthio) valerate], dioctadecyl tin di [4,4-bis (phenylthio) valerate], dioctyl tin di [4,4-bis (carboxyethylthio) valerate], butyl octyl tin di [4,4-bis (carbobutoxyethylthio) valerate].

dimethyl tin di [4,4-bis (dodecylthio) valerate], dioctyl tin di [ethyl 3.3-bis (carboxyethylthio) butyrate]

diphenyl tin di [3,3-bis (benzylthio)-3-phenyl propionate]. 20

Typical examples of compounds in Group III are

dioutyl tin 2-hydroxybenzaldi (2'-mercaptobenzoate), dibutyl tin propane-2,2-bis (2'-mercaptobenzoate)

diphenyl tin methane-bis (4-mercaptobenzoate), dilauryl tin diphenyl methane-bis (3-mercaptobenzoate), dimethyl in ethane-1,1-bis (2'-mercaptobenzoate).

Typical examples of compounds in Group IV are ditin di (dodecylthiomethyl thioacetate) [(C₄H₀)₂ OOCCH₂SCH₂SC₁₂H₂₅)₂], dilauryl tin di (methylthiomethyl-thiopropionate), dimethyl tin di (phenylthiomethyl-thioacetate), diphenyl tin di (benzylthiomethyl-thioacetate).

Typical examples of compounds in Group V are di (dibutyl tin) ethane 1,1,2,2-tetrakis (mercaptopropio-mate)

ci (dioctyl tin)-1-methyl ethane 1,1,2,2-tetrakis (mercaptoacetate), di (diphenyl tin) butane-1,1,4,4-tetrakis (mercaptopropionate), di (dilauryl tin) hexane-1,1,6,6-tetrakis (niercaptooctanoate).

Typical examples of compounds in Group A are bis (butyl tin) tri (propane-2,2-bis-β-mercaptopropionate); bis (butyl tin) tri (benzaldi-β-mercaptopropionate); bis (octyl tin) tri (propane-2,2-bis mercaptoacetate); bis (phenyl tin) tri (hydroxybenzaldi-β-mercaptopropionate); bis (methyl tin) tri (cyclohexyl-1,1-bis mercapto-octanoate); bis (benzyl tin) tri (2-hydroxy-4-methoxy-benzaldi mercaptobutyrate); bis (octadecyl tin) tri (2-butene-1,1-bis mercaptopropionate).

Typical examples of compounds in Group B are butyl tin tri [4,4-bis (dodecylthio) valerate], butyl tin tri [4,4-bis (phenylthio) valerate], octadecyl tin tri [4,4-bis (phenylthio) valerate], octyl tin tri [4,4-bis carboxyethylthio) valerate], butyl tin tri [4,4-bis (carbobutoxyethylthio) valerate], methyl tin tri [ethyl-3,3-bis (carboxyethylthio) butyrate], phenyl tin tri [3,3-bis (benzio)-3-phenylpropionate].

sical examples of compounds in Group C are bis ...yl tin) tri (benzaldi 2'-mercaptobenzoate), bis (butyl tin) tri propane-2,2-bis-2'-mercaptobenzoate, bis (phenyl tin) tri (2-hydroxybenzaldi 2'-mercaptobenzoate), bis (octyl tin) tri (methane-bis-4-mercaptobenzoate), bis (jauryl tin) tri (diphenylmethane bis (3-mercaptoben-

zoate), bis (methyl tin) tri (ethane-1,1-bis-2'-mercapto-benzoate).

Typical examples of comp unds in Group D are butyl till tri (dodecylthiomethyl thioacetate), lauryl tin tri chethylthiomethyl thiopropionate), methyl tin tri (phenylthiomethyl thioacetanoate), phenyl tin tri (benzylthiomethyl thioacetate).

Typical examples of compounds in Group E are bis (tributyl tin) propane-2,2-bis-β-mercaptopropionate, bis (tributyl tin) benzaldi-β-mercaptopropionate, bis (butyl octyl octadecyl tin) propane-2,2-bis mercaptoacetate, bis (trioctyl tin) propane-2,2-bis-mercaptooctanoate, bis (triphenyl tin) hydroxybenzaldi-β-mercaptopropionate, bis (trimethyl tin cyclohexyl-1,1-bis mercaptoacetate, bis (tribenzyl tin)-2-hydroxy-4-methoxybenzaldi mercaptobutyrate, bis (tri octadecyl tin) 2-butene-1,1-bis mercaptopropionate, tributyl tin propane-2,2-bis-β-mercaptopropionate, trioctyl tin benzaldi-mercaptoacetate

Typical examples of compounds in Group F are tributyl tin propane 2,2-bis mercaptosuccinate

bis (tributyl tin) propane 2,2-bis mercaptosuccinate, tris (hiphenyl tin) propane 2,2-bis mercaptosuccinate, tetra (trioctyl tin) propane 2,2-bis mercaptosuccinate.

Typical examples of compounds in Group G are triburyl tin 4,4-bis (dodecythio) valerate, tributyl tin 4,4-bis (phenylthio) valerate, trioctadecyl tin 4,4-bis (phenylthio) valerate, trioctyl tin 4,4-bis (carboxyethylthio) valerate, butyloctyl lauryl tin 4,4-bis (carbobutylethylthio) valerate, trimethyl tin 4,4-bis (dodecylthio) valerate, trimethyl tin 4,4-bis (dodecylthio) valerate, triphenyl tin 2,3-bis (carboxyethylthio) butyrate, triphenyl tin 3,3-bis (benzylthio)-3-phenyl propionate.

Typical examples of compounds in Group H are bis (tributyl tin) propane-2,2-bis (2'-mercaptobenzoate), bis (tributyl tin)-2-hydroxybenzaldi (2'-mercaptobenzoate), bis (triphenyl tin) benzaldi (2'-mercaptobenzoate), bis (trioctyl tin) methane-bis (4-mercaptobenzoate), bis (trilauryl tin) diphenyl methane bis (3-mercaptobenzoate), bis (trimethyl tin) ethane-1,1-bis (2'-mercaptobenzoate), tributyl tin propane-2,2-bis (2'-mercaptobenzoate)

Typical examples of compounds in Group J are tributyl the dodecylthiomethyl thioacetate, trilauryl tin methylthiomethyl thiopropionate, trimethyl tin phenylthiomethyl thiocatanoate, triphenyl tin benzylthiomethyl thioacetate.

Typical examples of compounds in Group K are trioctyl tin -1-methyl ethane 1,1,2,2-tetrakis (mercaptoacetone), tributyl tin ethane 1,1,2,2-tetrakis (mercapt proprociate), bis (tributyl tin) ethane 1,1,2,2-tetrakis (mercap opropionate), tris (tributyl tin) ethane 1,1,2,2-tetrakis (mercaptopropionate), tetra (tributyl tin) ethane 1,1,2,2tetrakis (mercapt propionate), tetra (triphenyl tin) butane 1,1,4,4-tetrakis (mercapt propionate), tetra (trilauryl tin) hexane-1,1,6,6-tetrakis (mercaptooctanoate).

(fauryl tin) tri (diphenylmethane bis (3-mercaptoben- 75 Via are benzaldi (mercaptoacetic acid), benzaldi (mer-

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captopropionic acid), 2-hydroxybenzaldi (n.c. capt acetic acid), 2-hydroxybenzaldi (mercaptopr pionic acid), 4hydroxy-3-methoxy-benzaldi (mercapt proponic acid), cyclohexyl-1,1-bis (mercaptoacetic acid), c; i hexyl-1,1bis (mercaptopropionic acid), methane-b mercaptoacetic acid), methane-bis (mercaptopropior acid), isodecane-1,1-bis (mercaptoacetic acid, isodecane-1, 1-bis (mercaptopropionic acid), propane-2, 2-bis (mercaptoacetic acid), propane-2,2-bis (mercaptopropionic acid), 4methylpentane-2,2-bis (mercaptoacetic acid), butane-2,2- 10 bis (mercaptoacetic acid), diphenylmethane-bis (mercaptopropionic acid), propene-3,3-bis (mercaptopropionic acid), 2-butene-1,1-bis (mercaptopropionic acid), propane-2,2-bis (mercaptosuccinic acid), 2-hydroxybenzaldi (2'-mercaptobenzoic acid), propane-2,2-bis (mercapto- 15 octanoic acid), phenylacetaldi (mercaptorcetic acid), propane-2-mercaptoacetic acid-2-mercaptop: pionic acid, and 2 - hydroxy - 4 - methoxybenzaldi (mercaptobutyric

Typical examples of compounds in Group VII are 4,4-bis (carbobutoxyethylthio) valeric acid, 4,4-bis (dodecylthio) valeric acid, 4,4-bis (phenylthio) valeric acid, 4,4-bis (carboxyethylthio) valeric acid, ethy,-3,3-bis (carboxyethylthio) butyrate

CU.—CCH,COOC.H.
CH,CH,COOH

3,3-bis (benzylthio)-3-phenylpropionic acid.

Typical examples of compounds in Group VIII are 2-hydroxybenzaldi (2'-mercaptobenzoic acid), propane-2,2 - bis (2'-mercaptobenzoic acid), methane-bis (4-mercaptobenzoic acid), methane-bis (3-mercaptobenzoic acid), ethane-1,1-bis (2'-mercaptobenzoic acid), phenylmethane bis (2-mercaptobenzoic acid).

Typical examples of compounds in Group 1X are dodecylthiomethyl mercaptoacetic acid, methylthic nethyl mercaptopropionic acid, phenylthiomethyl mercaptoacetic acid, butylthiomethyl mercaptoacetic acid, butylthiomethyl mercaptopropionic acid.

Typical examples of compounds in Group (are ethane-1,1,2,2-tetrakis (mercaptopropionic acid), I methylethane 1,1,2,2 - tetrakis (mercaptoacetic acid), per ane - 2,2,4,4-tetrakis (mercaptopropionic acid), ethane-1,2,2-tetrakis (mercaptoacetic acid), 1,1,4,4-tetrakis butane-(mercaptopropionic acid), hexane 1,1,6,6-tetrakis (mercapto-octanoic acid).

Example I

Butyl stannoic acid (0.91 mole) was heated in a vacuum (water pump) with 1.37 moles of propine-2,2-bis-β-mercaptopropionic acid and the water removed. The product was bis (butyl tin) tri (propane-2,2-bis-β-mercaptopropionate), a white solid melting or softening around 50-80° C., tin 21.5% (theory 21.6%), sulfur 17.68% (theory 17.4%). The solid appeared to be in polymeric form.

Example II

One mol of butyl tin triisopropylate was heated with 1.5 moles of propane 2,2-bis-β-mercaptopropionic acid in 200 ml. of isopropyl alcohol and there was recovered as the residue bis (butyl tin) tri (propane-2,7-1 is-β-mercaptopropionate) having the same pr perties as the product 65 of Example I.

Example III

0.0865 mol of butyl stannoic acid was here ed with 0.13 m 1 of benzaldi-β-mercapt propionic acid in a vacuum and the water removed to recover bis (butyl tin) tri (benzaldi-bis-β-mercaptopr pionate) as a pale yell w solid melting at 67-88° C., tin 19.5% (theory 19.05%), sulfur 15.82% (theory 15.42%). The solid appeared to be in polymeric f rm.

8 Example IV

0.15 m l f tributyl tin oxide was reacted with 0.15 mol of propane-2,2-bis-β-mercaptopropionic acid in 100 ml. f a mixture f equal amounts by v lume f hexane and benzene. The bis (tributyl tin) propane-2,2-bis-β-mercaptopropionate obtained was recrystallized fr m isopropanol as a white crystalline solid, M.P. 98-99° C., tin 28.0% (theory 28.6%), sulfur 7.52% (theory 7.73%).

Example V

0.15 mol of tributyl tin oxide was heated to reflux with 0.15 mol of benzaldi-β-mercaptopropionic acid in 100 ml. of a mixture of equal amounts by volume of hexane and benzene. The bis (tributyl tin) benzaldi-β-mercaptopropionate obtained was a white crystalline solid, M.P. 95-96° C., tin 26.9% (theory 27.0%), sulfur 7.38% (theory 7.30%).

Example VI

One mol of octyl stannoic acid and one mol of 4,4-bis (carboxyethylthio) valeric acid were heated in a vacuum and the water removed to recover octyltin tri [4,4-bis (carboxyethylthio) valerate].

Example VII

One mol of butyl stannoic acid was heated in a vacuum with 1.5 moles of benzaldi 2'-mercaptobenzoic acid and the water removed to recover bis (butyl tin) tri (benzaldi 30 2'-mercaptobenzoate).

Example VIII

One mol of butyl stannoic acid and one mol of dodecylthiomethylthioacetic acid were heated in a vacuum and the water removed to recover butyl tin tri (dodecylthiomethylthioacetate).

Example IX

One mol of tributyl tin oxide was heated to reflux with 2 moles of propane-2,2-bis (mercaptosuccinic acid) in 500 ml. of a mixture of equal amounts by volume of benzene and hexane to obtain tetra (tributyl tin) propane 2,2-bis mercaptosuccinate as the product.

When there was used only 0.5 mol of propane-2,2-bis (mercaptosuccinic acid) in this reaction the product obtained was tetra (butyl tin) propane-2,2-bis mercaptosuccinate.

Example X

One mol of trioctyl tin oxide was heated to reflux with 2 moles of 1 - methyl ethane-1,1,2,2-tetrakis (mercaptoacetic acid) in 750 ml. of a mixture of equal parts by volume of benzene and hexane to obtain trioctyl tin-1-methylethane 1,1,2,2-tetrakis (mercaptoacetate) as the product.

Example XI

One mol of tributyl tin oxide was heated to reflux with 0.5 mol of ethane-1,1,2,2,-tetrakis (mercaptopropionic acid) in 500 ml. of a mixture of equal parts by volume of benzene and hexane to obtain tetra (tributyl tin) ethane 1,1,2,2-tetrakis (mercaptopropionate) as the product

Example XII

One mol of tributyl tin oxide was heated to reflux with 2 m les f 4,4-bis (dodecylthio) valeric acid in 750 ml. of a mixture f equal parts by volume of benzene and hexane to obtain tributyl tin 4,4-bis (dodecylthio) valerate as the product.

Example XIII

The procedure of Example XII was repeated replacing the 4.4-bis (dodecylthi) valeric acid by 2 moles f dodecylthiomethylthioacetic acid to produce tributyl tin dodecylthiomethylthioacetate as the product.

Example XIV

e mol of tributyl tin oxide was heated to reflux with m l of propane-2,2-bis (2'-mercaptobenz ic acid) in 10 ml. of a mixture of equal parts by v lume f benzene and hexane to obtain bis (tributyl tin) propane-2,2,-bis ("-mercaptobenzoate) as the final product.

The stabilizers of the present invention can be used with halogen containing vinyl and vinylidene resins in which the halogen is attached directly to a carbon atom in the polymer chain. Preferably, the resin is a vinyl halide resin, specifically, a vinyl chloride resin. Usually, the vinyl chloride atone or a mixture of monomers consisting of vinyl chloride alone or a mixture of monomers comprising at least 70% vinyl chloride by weight. When vinyl chloride copolymers are stabilized, preferably the copolymer of vinyl chloride with an ethylenically unsaturated compound copolymerizable therewith contains at least 10% of polymerized vinyl chloride.

As the chlorinated resin there can be employed chlorinated polyethylene having 14 to 75%, e.g., 27% chlorine by weight, polyvinyl chloride, polyvinylidene chloride, polyvinyl bromide, polyvinyl fluoride, copolymers of vinyl chloride with 1 to 90%, preferably 1 to 30%, of a copolymerizable ethylenically unsaturated material such as vinyl acetate. vinyl butyrate, vinyl benzoate, vinylidene chloride, diethyl fumarate, diethyl maleate, other alkyl furnarates and maleates, vinyl propionate, methyl acrylate, 2-ethylhexyl acrylate, butyl acrylate and other alkyl acrylates, methyl methacrylate, ethyl methacrylate, butyl methacrylate and other alkyl methacrylates, methyl alpha chloroacrylate, styrene, trichloroethylene, vinyl ethers such as vinyl ethyl ether, vinyl chloroethyl ether and vinyl phenyy ether, vinyl ketones such as vinyl methyl ketone and vinyl phenyl ketone, 1-fluoro-1-chloroethylene, acryl-

te, chloracrylonitrile, allylidene diacetate and chloroene diacetate. Typical copolymers include vinyl chloride-vinyl acetate (96:4 sold commercial as VYNW), vinyl chloride-vinylacetate (87:13), vinyl chloride-vinyl VI. VIa, VII, VIII, IX and X, there is normally used 0.01 to 10% of the tin compound by weight of the resin and the thioacetal or thi ketal is also normally used in an amount of 0.01 to 10% by weight of the resin. More 1 clerably, 0.2 to 5% of the tin compound and 0.2 t 5%

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in the thioacetals or thioketal is employed based on the wight of the resin.

The following examples employ the stabilizers of the present invention with vinyl chloride resins. Without exception there was greater stabilization than when the equivalent amount of dibutyl tin dilaurate was employed. Many of the compounds and compositions were superior to dibutyl tin bis (isooctylthioglycolate) as stabilizers for vinyl chloride resins.

The following examples (except comparison Examples 1 and 2) illustrate the stabilizing effect of the additives of the present invention. It should be noted that the first yellowing does not necessarily limit the usefulness of the stabilizer. The stabilizer tests were carried out at 360° 20 1°. (182° C.) in the conventional forced draft oven. In the tables 101 EP and 103 EP designate Geon 101 EP and G. on 103 EP which are vinyl chloride homopolymers manufactured by B. F. Goodrich. VYNW designates a 9c.4 vinyl chloride-vinyl acetate copolymer and St. Gobain, a commercially available vinyl chloride resin. In the tests all parts are expressed as parts per 100 parts by weight of the resin. The letter Y designates the time in minutes at which color first appeared. The letter B designates the time in minutes at which the resin became

Example 1
[Dibutyl tin dissurate—Molecular weight 525]

35 Ports stabilizer. Resid Dioctyl phthalate (parts) Rel. its	0. 95	0.94	1.88	2.82
	103EP	101EP	103EP	103EP
	0	50	50	0
	Y-0	Y-15	Y-30	Y-0
	B-60	B-45	B-75	B-76

Example 2
[Dible.... tin (isooctylthioglycolate)—Molecular weight 580]

Parts stabilizerResin	0. 94 103EP	0. 775 103E P	0.95 1 03E P	1.0 101EP	1.67 VYNW	1. 11 St. Go- bain
Dioctyl phthalate (14218). Results	50 Y-45 B-90	50 Y-45 B-75	Y-45 B-60	У-30 В-75	Y-78 B-75	50 Y-60 B-75

acetate-maleic anhydride (86:13:1), vinyl chloridevinylidene chloride (95:5), vinyl chloride-diethyl fumatate (95:5), vinyl chloride-trichloroethylene (95:5), vinyl chloride-2- ethylhexyl acrylate (80:20).

The stabilizers of the present invention can be incorperated with the resin by admixing in an appropriate mill or mixer or by any of the other well-known methods which provide for uniform distribution throughout the resin compositions. Thus, mixing can be accomplished by milling on rolls at 100-160° C.

In addition to the novel stabilizers there can also be incorporated with the resin conventional additives such as plasticizers, pigments, fillers, dyes, ultraviolet light absorbing agents, densifying agents and the like.

If a plasticizer is employed, it is used in conventional amount, e.g., 30 to 150 parts per 100 parts f resin. Typical plasticizers are di-2-ethylbexyl phthalate, dibutyl sebacate, dioctyl sebacate, tricresyl phosphate.

The tin containing stabilizers in Groups I, Ia, II, III, IV, B, C. D. E, F, G. H, J and K are normally used amount of 0.01 to 10% by weight f the chlorinated resin. More preferably, 0.2 to 5% f the tin compound is used by weight of the resin.

When a mixture of dihydrocarbon tin oxide or sulfide or trihydrocarbon tin oxide or monohydrocarbon stannoic acid is employed with the thioacetals or ketals f Groups 75

Example 3

[Dibutyl tin isodecane-1,1-bis (mercaptopropionate)—Molecular weight 580]

		
	Part. stabilizer	0.86
	Resin	103EP
5	Results.	Y-45
-		B-75

Example 4

[i.bibutyl tin benzaldi (mercaptopropionate)—Molecular weight 580]

•		
	Paris stabilizer	1.60
	Reciti	103EP
	Direct v1 phthalate (perts)	90 Y-75
		B-120
=		ļ

Example 5

[Di c'yl tin propane-2,2-bis (mercaptopropionate)—Molecular weight 594]

0	Parts stabilizer Epox'dized soya dil Epcadized isooctyl aleate Resin. Resin.	0	0. 453 0 0. 453 101EP Y-75 B-60	0. 453 6 0. 453 101EP Y-75 D-185
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Example 6

		Exam											
[Dibuty]	tin bensale	li (mercapt	oacetato)-	Molecular	weight 50	2]		-					
Parts stabilizer Resin Dioctyl plthalate (pa Tricresyl phosphate		.75	.5 P 103	0 1031	.75 1 EP 103	68 EP 50	1. 4 103EF	9 0 0					
(parts) Results						-80 -165	B-16						
		!						-	F-	ample 1	7 9		
•		ample 7		foleenlar	-oleht			low nomad :		-		onic acid)—	
[Dibutyl tin propane	-2,2-bis (m	ercaptoproj 82]	olonate)—:	1010001181	werkn.		ĮC	ombonia:	Molecu	ier weigh	224]		
Parts stabilizer Epoxidized soyn oil	0.715 0 103EP 50 Y-90	0 Y-75	0 Y-75	0.715 0 0.1EP 0 V-75	0.905 1 103EP 0 Y-75 B-75	D A R D	esin	ozide pound thalate			0.87 0.19 103EP 50 Y30 B-75	0.87 0.25 103EP 0 Y-0 13-75	0. 37 0. 34 101EP 0 Y-0 B-75
7.030.00	B-105	B-75	B-120	1:-90	2	0 -				1		<u> </u>	
					Exam								
		(Comp	ound: Ber	reuldi bis ((mercapto	propi	ionie acid)	Molecul	ar weight	800]	<u> </u>		
Dibutyl tin oxido (p Abovo compound (p Resin	rts) (parts)	103ÆP	0.74 0.45 103EP 50 9 Y-90 B-125		0. 4/ 103EF 50 Y-12/	000	0.74 0.88 103EP 0 50 Y-60 B-165	0.74 0.49 103EP 50 0 Y-75 B-155	0. 37 0. 25 103EP 0 0 Y-15 B-105	0.37 0.30 101EP 0 0 Y-80 B-75	0. 87 0. 46 101EP 0 0 Y-80 B-78	0.37 0.23 101EP 0 0 Y-60 B-75	0.87 0.46 101EP 50 0 Y-80 B-90
		•			Exan		. 0			•			
		Compos	nd: O-hyr	ro-vbenst		•		acid)Mol	ecular wei	ght 316)			•
		Compos	· · · · · · · · · · · · · · · · · · ·		- 1		T	0.87	0.87	0, 87	0.37 0	855 0.87	0. 37
Dibutyl tin oxide Above compound Resin Dioctyl phthalate Tricresyl phosphate		103 EP 50	0.74 0.59 103 EP 50	EP 0	St. Jobain 50	0, 37 1, 275 103 EP 50	103 EP 0	0.30 St. Gobain 0 Y-60	0.32 101 EP 0 V-80	0. \$5 103 EP 0 0 Y-15	0.50 103 EP 0	0.77 0.475 101 101 EP EP 0 0 0 0 7-15 Y-30	0.44 101 EP 50 0 Y-75
Results		Y-90 B-135	Y-90 B-165	Y 60 R-75		Y-0 0 3-120		B-105	B-78	B-90		-120 B-90	B-106
*Indicates resin	was not ve	ry dark wh	en test was	: crminat	ed after 16	 ئند ڈ	nutes.						
-Indicates rom		cample I						•		xample			
10	-	o-1,1-bis(m		lie a cid) —	•	4 5		[Compot	md: Meth: Mole	ane-bis (m cular weig	ercaptoac ht 196]	etic acid)—	
(Compoun	Molec	ular weight	322]				Dibutyl	tin oxide				0. 87 0. 20	•
Dibutyl tin oxido				0. 37 0. 50			Resin Results	m pound				103EP Y-15	
Resin				101EI Y-0	•	50	Deptition.					B-60	
Results				B-90				[Compou	nd: Butar	Example 10-2,2-bis (10-2,2-bis wei	mercapto	acetic acid)—	
	E	xample 1	11			5 5	Dibntyl	tin oxide.				0.87	0.87
[Compou	nd · Pronan	o-2,2-bis (m cular weigh	erceptoaco	ntic icid)—			Above o	ompound				0. 21 103EP Y-80 B-60	0. 40 101EP Y-15 B-60
Dibutyl tin oxide Above compound Resin	IUSEP	0. 87 0. 22 St. Gobain	0.37 0.36 101EP	0. 37 0. 20 1::1EP Y-15	0. 37 0. 21 103EP Y-30	60		[Compour	nd: Cyclob	Example exyl-1,1-b lecular we	is (merca)	ptoacetic scid)	<u>.</u> .
Results	Y-30 B-105	Y-15 B-90	Y-30 B-60	B-60	B-60		Dibuty	tin oxide.		0.3		87 0.57	0. 87 0. 40
	<u>'</u>	Example	12			65	Above of Resin Dioctyl	ompound phthalate		103E	0 Y-	O Y-30	101EP 50 Y-30 B-75
[Compound:	Dinhenvim	othane-bis	(mercapto)	pr. pi onic (aciá)—					Example	e 17	!	
	Mol	ocular welg	us ostoj			- 70	Comp	ound: 4,4-	bis (dodec			-Molecular v	reight 490]
Dibutyl tin oxide	0.87	0.87 0.85	0. 87 0. 35	0. 37 0. 57	0.8	77 ·	·	-1 etn aufde				0.87	0. 87 0. 80
Above compound Resin Dioctyl phthalate	103EP	101EP	103EP 0 Y-30	101EP 0 Y-30		50	Above	compound	l			103EP Y-0	103EP Y-0
Results	Y-30 13-90		B-75	B-75				J				B-60	B-76

Example 18						Example 26		
(Compound: Isodecane-1,1-bis (mercaptopropionic acid)— biolecular weight 250]						(f. unpound: cyclohaxyl-1,1-bis (mercaptopropionic acid)—h weight 292]	Molecular	
Dibutyl tin oxide		0.37 0.315 103E1' 0 Y-60 B-75	0. 37 0. 53 101 EP 0 Y-30 B-75	0. 37 0. 53 101 EP 50 Y-30 B-75		utyl tin oxide	0. 37 0. 45 101EP 50 Y-75 B-105	
	cample 19				10	Example 27		
1Compound: 4-bydroxy-3-meth Moleculi	oxybenzeldi ir weight 810		ptopropion	ic scid)—		(Compound: Propone-2,2-bis (mercaptosuccinic scid)—Kolet weight 340)	cular	
Pributyl tin oxide			0. 37 0. 36 103 E P Y-0 B-75	0. 37 0. 53 101 E P Y-0 13-75		Dibutyl tin oxide	0. 37 0. 51 103 E.P 50	
Ex	cample 20			-		1. sults	Y-45 B-115	
Compound: Propane-2,2-bls wei	(mercaptop: ght 252j	ropionic	acid)—Mo	locular	20	Example 28		
O Dioctyl tin oxide O S4 Above compound 0. 54 Resin 103EP	0. 37 0 0. 38 101 E P	0.37 0 0.20 101 E P	0.87 0 0.48 101EP	0. 37 0 0. 22 103 E P	25	[Compound: Propane-3,3-bis (mercaptopropionic acid)—M weight 250]	folecular	
Dioctyl phthainte 0 Results Y-60 B-75	0 Y-50 B-90	Y-60 B-90	Y-75 B-105	50 Y-103 B-120		Dil utyl tin oride	0, 37 0, 20 101 E P Y-0	
Ex	ample 21				30	B-75	B-76	
[Compound: 4,4-bis (phenylth	lo) valeric a	cid—Mo	lecular wei	ight 318]		Example 29		
Dibutyl tin oxide			0. 37 0. 48 101 B P Y-0 B-60	0. 37 0. 25 101 E P Y-0 B-60	35	(Compound: 2-butene-1,1-bis (mercaptopropionic acid)—Mole weight 264) Dibutyl tin oxide 0.87	0.37	
[Compound: 4.4-bis (earbobut	cample 22 oxy-ethylthiveight 422	io) vale	ric acid—3	Motecular	40	Above compound		
ibutyl tin oxide		0. 37 0. 66	0. 37 0. 33	0. 37 0. 66	10	Example 30		
Pesin Pioetyl phthalate Results		0 Y-60 B-90	101EP 0 Y-30 B-75	103EP 50 Y-60 B-90	45	[Compound: Ethyl-3,3-bis (carboxy-ethylthio) butyrate—Mole weight 324]	ecular	
Ex- Compound: 4,4-bis (earboxy-et)	ample 23 bylthio) valo 287]			·	***	Dibutyl tin oxide	0. 37 0. 49 103 R P 50 Y-75 B-90	
Dibutyl tin oxide	0. 37 0. 34 103EP	0. 87 0. 32 10LEP	0. 37 0. 18 101EP	0. 37 0. 32 103EP	50	Example 31		
Dicetyl phthalate	Y-75 B-103	0 Y-30 B-75	Y-80 B-75	50 Y-60 B-90	55	[Compound: Pentano-2,2,4,4-tetra (mercaptopropionic acid)—M weight 489]	lolecular	
Fy	ample 24	!			ออ	F. mula: HOOCCH, CH, S SCH, CH, COOH		
Compound: 2-hydroxybenzaldi	•	tobenzo	ic acid)—	Molecular		нооссилсия есилсилсоон		
Above compound Above compound Action	0.37 0.42 103EP 50 Y-30	0.37 0.37 101EP 0 Y-0	0.37 0.70 101E1' 0 Y-0	0. 37 0. 70 103EP 50 Y-15	60	Identify	0. 37 0. 39 101EP Y-80	
	B-105	B-60	B-00	B-60	65	B-75	B-75	
Compound: Ethane-1,1,2,2-tetra	cample 25 akls (mercap reight 456)	topropic	nic acid)—	Molecular		Example 32	cular	
cyl tin oxide			0.3 0.2 103EI 5 Y-1: B-7:	5 0 5	70	Weight 232	0.40 0.22 101BP 50 Y-60	
		_	D-1		75	B-78	B-106	

15

16 Example 40

	Example 33						Example 40		
Compound: Benzaldi (m		cid)—Mol	e tilar welf	tht 300)		(Compound	d: Methane-bis-(mercaptosce	tic acid)]	
ibutyl tin sulfide			0.40	0.40		Above compound		0.44	0.44
			e 26	0. 26 101E P	5	Rutyl stannoic scid		0.81	0.46
			50	0		Resin		101EP	101EP
esults				Y-15 B-75		Dioctyl phthelate		Y-80 P-75	Y-15 B-60
<u></u>	Example 34	!_			10			1,-1,0	
(Compound: cycl	ohexyl-1,1-bis (me Molecular waigh	reaptopro	piralo sci dj	-	10		Example 41		
						[Compound:	Indecane-1,1-bis-(mercaptos	cetic acid)]	
ibutyl tin sulfide oove compound			0. 40 0. 28	0. 40 0. 28		A born compound		0.72	0. 72
esinioctyl phthalate			151EP	101BP 50	15	Butyl stannoic acid		0.31	· 0 0. 46
mults			45	Y-60 B-105		Tributyl un oxide		101BP	10LEP
			1 75	D-100		Results		Y-0 11-45	Y-0 B-80
The term thioace	tal as used h	erein is	i. acco	rdance				<u> </u>	
ith accepted termin	nology generic	to thiol	ko als.		20		Example 42		
	Example 35 Bis (butyl tip) t		me.a.			[Compound:	: 4,4-bis-(carboxyethyithio) v	aleric acid)	
[Compound mcres	ptopropionate)—h	fol. wt. 110	0.				, <u>.</u>	0,64	0. 64
	815 0.815	0.815	U. S15	0, 408	25	Butyl stannoic scid	, an appartunes, pupe, cano, cano, cano, c	0.81	. 0
bove compound 0. csin 103	815 0.815 EP 103EP	103EP	10: SP	101EP		Tributyi tin ozide		101EP	0. 46 101 E P
ioctyl phthalate_	50 50 -45 Y-45	Y-45	7:-30	Y-30		Results		Y-30	Y-0
	1-75 B-90	B-00	ी-46	B-45				B-45	B-60
						,	•		
		mple 36							
[Compound: Bis (but)	yl tin) tri (benzald	ii-bis-6-mc	respontor	ionato)—	-Mol	wt. 1244.2]			
bove compound	0.925	. 463		0. 925	0.9	25 0.925	•		
lesio	101EP 10	IEP 10	ner Vi	WMY	101 E	P 101EP			
Dioctyl phthalate Spoxidized soybean oil.		ŏ	0	Ō		5 0			
0-di tert, butyl p-cresol		Y-30	- 18	Y-30	v.	0 0.5 -15 Y-80			
lesuits			15-45	B-45	B-1				
									
							E		
	Example 3				45		Example 43 and: 4,4-bis-(dodecy)thio) val	eric scidi	
[Compoun	d: Bis (tributy) ti ptopropionate—bi	n) propan ol. wt. 829.	ε-2.:⊹bis-β- .4)					T	
					-	Above compound		0,31	1. 1
bove compound		l	0,62	0.3	ı	Telbatal tip azida		. 0 1	0.40
Resin.			101EP	101EI	^ - <i>/</i>	Resin		101EP	101EF
Results			13-60	B-8) Results		B-45	B-4
	 -				•				
								•	
		mple 38			3/-1	W+ 10471			
(Compound: Bis-	(tributy) tlp) ben	zaldi-8-me	respioproj	- OUBLIDE	HUI.	1			
Above compound		0.33	u. 66 101 E.P	0.66 101EP		0.66 0.66 EP VYNW			
ResinDioctyl phthalate		0	.0	0		0 0			
Epoxidized soybean oil		0	0	5		0			
O.O.di tert. butyl p- cresol	0	0	0	_ 0		0.5			•
Results	Y-0 B-75	Y-0 B-80	۲-15 4-75	Y-0 B-75		Y-0 3-60 Y-15 B-30			
									
	E	20					Example 44		
- - -	Example .		ntavatie e	eid)		Compound	1: 2-butens-1,1-bis-(mercapto	propionie sc	ld)]
(Compound	: Cycloneryi-1,1-l	VIS (HICTOR)	Τ	T					T
Whose combound			0.89	0.		70 Above compound		0.81	
			0.81	0.	. 48	Tributy! tin onice.) 0.
Tributyl tin exide			101RP	1011	2P	Resin		101E) l ·
Dioctyl Dribanic			Y-30		50 -15	Besults		1 17)
Results			B-60		-00			B-60	' "
						7K			

Example 45

'ompound.	Ethanc-1.1.2.2-te	trakis_(mercan	topropionic neid)]

A bove compound Itutyl stannoic acid Prioutyl tin oxide Iresin I extyl phthalate I esults	0.81 0 101EP	1.0 0.46 101E1' 60 Y-0 B-45
	B-00	B-45

Example 46

[Compound: 2-hydroxybensaldi (2'-mercaptobensoid acid)]

Example 47

[Compound: Methane-bis-(mercaptopropionic acid)]

A nove compound But vi stannoic acid Tributyl tin oxide Trisic Union phthalate Results	101EP	0. 56 0 0. 46 101EP 0 Y-0 B-75
{	35-60	B-75

Example 48

[Compound: Benzaldi (mercaptopropionic scid)]

Anove compound Unityl stanuoic acid Privatyl tin oxide P	0.67 0.81 0 01EP Y-30 B-60	0. 67 0. 46 101EP Y-0 B-75
---	---	--

The tin containing mercaptoacetal (including mercaptoketal) compounds of the present invention are not only 40 useful for stabilizing halogen containing resins but are also useful in stabilizing polymers of monoolefins having 2 to 4 carbon atoms. While the stabilizers of the present invention can be used with polyethylene, polypropylene, cthylene-propylene copolymers (e.g., a 50-50 copolymer), 45 polybutylene and polyisobutylene, they are preferably employed with polymers and copolymers of polypropylene.

The present invention is suitable for the stabilization of the monoolefin polymers regardless of the method employed to prepare the polymer. Thus, there can be 50 stabilized polyethylene, polypropylene, polybutylene and copolymers of ethylene with propylene prepared with Ziegler type polymerization catalysts. e.g., trialkyl alumiraum (tributyl aluminum) with titanium tetrachloride or dibutyl beryllium with titanium tetrachloride. The poly- 55 mers can be prepared using any of the Ziegler type of catalysts as set fortth in Salyer Patent 2.985,617, issued May 23, 1961, for example. However, the stabilizers of the present invention can be employed with polymers of monoolefins prepared by other processes, e.g., polyethyl- 60 one prepared under high pressure as set forth in Fawcett Fatent 2,153,553, for example, or polyethylene, polypropylene or copolymers prepared using Phillips Petroleum or Standard Oil of Indiana type catalysts.

The tin containing mercaptoacetals of the present in- 65 vention can be used as stabilizers in an amount of 0.01 to 10% by weight of the monoolefin polymer. Preferably, 0.1 to 5% of the stabilizer is employed. When the otin mercaptoacetals are employed t gether with stabilizers, usually 0.01 to 10%, and preferably 70 6.. 10 5%, of total stabilizer based on the weight of tile polymer is employed.

While the organotin mercaptoacetals can be employed atione, synergistic action has been observed when they

ticularly go d results are obtained when there is employed in additi n to the organ tin compound a neutral sulfur c mpound having a thio linkage beta to a carbon at m having both a hydrogen atom and a carboxyl group attached thereto. Such compounds are used in an amount ... 0.01 to 10% by weight, preferably 0.1 to 5%. The preferred thio compound is dilauryl thiodipropionate. Other thio compounds include

distearyl-3,3'-thiodipropionate (dioctadecyl-thiodipropionate).

dicyclohexyl-3,3'-thiodipropionate, dicetyl-3,3'-thiodipropionate, dihexyl-3,3'-thiodipropionate, dicatyl-3,3'-thiodipropionate, divenzyl-3,3'-thiodipropionate, lauryl myristyl-3,3'-thiodipropionate, diphenyl-3,3'-thiodipropionate, di-p-methoxyphenyl-3,3'-thiodipropionate. didecyl-3,3'-thiodipropionate, dibenzyl-3,3'-thiodipropionate, diethyl-3,3'-thiodipropionate,

lauryl ester of 3-methylmercapto propionic acid, lauryl ester of 3-butylmercapto propionic acid, 25 lawyl ester of 3-laurylmercaptopropionic acid. phenyl ester of 3-octylmercapto propionic acid, lauryl ester of 3-laurylmercapto propionic acid.

lauryl ester of 3-phenylmercapto propionic acid, Luryl ester of 3-benzylmercapto propionic acid, 30 lauryl ester of 3- (p-methoxy) phenylmercapto propionic

acid, lauryl ester of 3-cyclohexylmercapto propionic acid, lauryl ester of 3-hydroxymethylmercapto propionic acid, my istyl ester of 3-hydroxyethylmercapto propionic acid,

octyl ester of 3-methoxymethylmercapto propionic acid, dilauryl ester of 3-carboxymethylmercapto propionic

dil..uryl ester of 3-carboxypropylmercapto propionic acid, dilauryl-4,7-dithiasebactate,

dilauryl-4,7,8,11-tetrathiatetradecandioate.

dimyristyl-4,11-dithiatetradecandioate. lateryl-3-benzothiazylmercaptopropionate.

as well as other alkyl, cycloalkyl and aryl esters of the buta thiocarboxylic acids set forth in Gribbins patent 2.519,755. Preferably, the esterifying alcohol has 10 to

Other beta thiocarboxylic acids include stearyl (1,2dicarboethoxyethylthio) acetate, stearyl (1,2-dicarbolauryloxyethylthio) acetate, lauryl (1,2-dicarboethoxyethylthio) acetate or the like. Compounds of this type can be made in known fashion by addition of an alkyl ester of mercaptoacetic acid to a dialkyl ester of maleic acid. Similar beta thiocarboxyl compounds can be used which are made by addition of an RSH compound across the maleic ester double bond and where R is alkyl, aryl, aisylcarboxyalkyl, arylcarboxyalkyl, or aralkyl. Examries of such compounds are decylthiodilaurylmaleate, phenylthiodioctyl maleate, cetyl (1,2-dicarboethoxyethylthio) propionate and benzylthiodimyristyl maleate.

Similarly, useful beta thiocarboxyl compounds can be prepared by addition of the RSH compounds as defined above across the double bond of dialkyl itaconates, dialkyl citraconates, dialkyl fumarates, or trialkyl aconitales, e.g., the addition product of lauryl mercaptan with dibutyl itaconate, the addition product f the stearyl ester of mercaptoacetic acid with dilauryl itaconate, the addition product f butyl mercaptan with dilauryl citraconate, the addition product of lauryl mercaptan with tributyl aconitate, the addition product of the lauryl ester of mercapto propionic acid with triethyl aconitate.

The thermal stability of the polypropylene and other polymers of a monoolefin is adversely affected by impurities including residual catalyst. When thermal staare employed together with certain other stabilizers. Par- 75 bility is important in addition to oxidative stability, it

60

tate)

has been found valuable to include alkaling earth metal salts of fatty acids in an amount of 0.0; to 10% by weight, preferably 0.1 to 5%, in the tin salt of a carb xy mercaptal formulations. Examples of such salts are calcium stearate, calcium 2-ethylhexoate, calcium myristate, calcium leate, calcium ricinoleate, calcium myristate, calcium palmitate, calcium laurate, barium laurate, barium stearate and magnesium stearate. Other fatty acid salts such as cadmium 2-ethylhexoate, zinc stearate, and cadmium stearate can also be employed.

Particularly effective synergistic stabilizing compositions have been obtained by utilizing a nexture of (1) the organotin compound, (2) the thio compound, particularly dilauryl thiodipropionate, and (3) the alkaline earth metal salt of a fatty acid.

The addition of phenolic antioxidants ii. an amount of 0.01 to 10% by weight, preferably 0.1 to 5%, also has proved effective. Examples of such piccols include 2,6-di-t-butyl-p-cresol, butylated hydroxyanisole, propyl gallate, 4,4'-thiobis (6-tertiary-butyl-m-cre.ed), 4,4'- cy- 20 clohexylidene diphenol, 2,5-di-tertiary-an i hydroquinone, 4,4'-butylidene bis (6-tertiary-butyl-n-cresol), hydroquinone monobenzyl ether, 2,2' - meth, i:ne - bis (4methyl-6-t-butylphenol), as well as the other phenols set forth in Salyer Patent 2,985,617. Omer suitable 25 phenols include 2-tertiary-butyl-4-decyloxyphenol, 2-tertiary-butyl-4-dodecyloxyphenol, 2 - tertiary - butyl - 4octadecyloxyphenol, 4,4'-methylene - bis - (2,5 - ditertiary butyl phenol), p-aminophenol, N-lauryl-11-aminophenol, 4.4' - thiobis (3 - methyl - 6 - t - butyl - ρ benol), bis [o-(1,1,3,3-tetramethylbutyl) phenol] sulfide, 4-acetyl- β resorcyclic acid, A-stage p-tertiary butylyhenol-formaldehyde resin, 4-dodecyloxy-2-hydroxy-ben ophenone, 3hydroxy-4-(phenylcarbonyl) phenyl palmii te, n-dodecyl ester of 3-hydroxy-4-(phenylcarbonyl) phenyxyacetic acid and t-butylphenol.

The use of epoxy compounds in an amount of 0.01 to 10% by weight, preferably 0.1 to 5%, in the organotin compound formulations has also been found valuable. Examples of such epoxy compounds include epoxidized soya oil, epoxidized lard oil, epoxidized oli e oil, epoxidized linseed oil, epoxidized castor oil, epoxidized peanut oil, epoxidized corn oil, epoxidized triag oil, epoxidized cottonseed oil, epichlorhydrin bisph : ol A resins, phenoxy-propylene oxide, butoxy-propylene oxide, epoxidized neopentylene oleate, glycidyl epoxymarate, epoxidized a-olefins, epoxidized glycidyl soyate, dicyclopentadiene dioxide, cpoxidized butyl tallate, styrene oxide, dipentene dioxide, glycidol, vinyl cyclohenene dioxide, glycidyl ether of resorcinol, glycidyl ether of hydroquinone, glycidyl ether of 1,5-dihydrox, naphthalene, epoxidized linseed oil fatty acids, allyl phycidyl ether, butyl glycidyl ether, cyclohexane oxide, -!-(2,3-epoxypropoxy) acetyl phenone, mesityl oxide eroxide, 2-ethyl-3-propyl glycidamide, glycidyl ethers of glycerine, pentaerythritol and sorbitol, and 3,4-epoxycyclohexane-1,1dimethanol bis (9,10-epoxystearate).

The percent of stabilizer in the following examples is based on the amount of polymer being 100%.

Example 49

0.5% of the indicated organotin merc....tal salt was blended into polypropylene having an initial melt index at 190° C. of 0.8. The formulation was thoricated into 65 a 20 mil thick strip and heated in an oven at 140° C. The unstabilized polypropylene degraded after 4 hours under these conditi ns.

	Time to Degrada-	
Stabilizer: tion	f Polymer (hours) 7	0
Dibutyl tin propane-2,2-bis (mercar		
Dibutyl tin propane-2,2-bis (mercar	otoprogionate) _ 48	
Dincopentyl tin pr pane-2,2-bis	mercar topropio-	
nate)		
Dioctyl tin propane-2,2-bis (mercag	ptopropionate) _ 24 7	5

Time to Degradation f P lymer (hours) Stabilizer: Dibutyl tin butane-2,2-bis (mercaptoacetate) .___ 72 Dibutyl tin butane-2,2-bis (mercaptopropionate) _ 24 Dibutyl tin 2-ethylbutane-1,1-bis (mercapto propio-______ 24 nate)_ Dibutyl tin isodecane-1,1-bis (mercaptopropionate). Dibutyl tin benzaldi (mercaptoacetate) _____ 72 Dibutyl tin benzaldi (mercaptopropionate) ____ 24 Dioctyl tin benzaldi (mercaptopropionate) ____ 48 Dibutyl tin o-hydroxybenzaldi (mercaptoacetate) _ 72 Dibutyl tin o-hydroxybenzaldi (mercaptopropionate) _____ Dioctyl tin o-hydroxybenzaldi (mercaptopropio-._____ 48 nate) ___. bis (Tributyl tin) propane-2,2-bis (mercaptopropionate) _. bis (Tributyl tin) benzaldi (mercaptopropionate) _ 48 bis (Butyl tin) tris [propane-2,2-bis (mercaptopropionate)] ____ bis (Butyl tin) tris [benzaldi (mercaptopropionate)] __ Dioctyl tin o-hydroxybenzaldi (mercaptoacetate) _ 48 Dioctyl tin isodecane-1,1-bis (mercaptoacctate) ___ 48 Dioctyl tin isodecane-1,1-bis (mercaptopropio-Dibutyl tin isodecane-1,1-bis (mercaptoacetate) __ 48 Dibutyl tin 1-carboethoxypropane-2,2-bis (mercaptopropionate) ____ Example 50

A formulation consisting of 0.166% of the indicated organotin mercaptal salt, 0.166% of dilauryl thiodipropionate and 0.166% calcium stearate was blended into the same polypropylene as employed in Example 49 and the strips obtained heated to 140° C. The time to degrade the polymer using the indicated tin compound in this formulation was as follows:

Organotin compound (with dilaurylthiodipropionate and calcium stearate): Time (hours) Dibutyl tin propane-2,2-bis (mercaptoacetate) Dibutyl tin propane-2,2-bis (mercaptopropionate) _____ Dineopentyl tin propane-2,2-bis (mercaptopropionate) _____ Dibutyl tin butane-2,2-bis (mercaptoacetate) __ 96 Dibutyl tin isodecane-1,1-bis (mercaptopropionate) _____ 96 Dibutyl tin benzaldi (mercaptoacetate) ____ 120 Dibutyl tin benzaldi (mercaptopropionate) ___ 24 Dioctyl tin benzaldi (mercaptopropionate) ___ Dibutyl tin o-hydroxybenzaldi (mercaptoacetate) . 120 Dibutyl tin o-hydroxybenzaldi (mercaptopropio-240 nate) __. bis (Tributyl tin) propane-2,2-bis (mercaptopropionate) _. bis (Tributyl tin) benzaldi (mercaptopropionate) _. bis (Butyl tin) tris [propane-2,2-bis (mercaptopropionate)] _____ bis (Butyl tin) tris [benzaldi (mercaptopropionate)] ___. 48 Dioctyl tin o-hydroxybenzaldi (mercaptoacetate) _____ 168 Dioctyl tin o-hydr xybenzaldi (mercaptopropionate) _____ Dioctyl tin isodecane-1,1-bis (mercaptoacetate) _ 216 Dioctyl tin isodecane-1,1-bis (mercaptopropionate) _____

Dibutyl tin isodecane-1,1-bis (mercapt ace-

Dibutyl tin 1-carboeth xypropane-2,2-bis (mercapt propionate)

Example 51

formulation consisting of 0.1% f the indicated continuous mercaptal salt, 0.1% dilauryl thiodipropi nate, 0.1% 2,6-di tertiary butyl p-cresol and 0.1% stearyl mercaptoacetate was blended into the same polypropylene as employed in Example 49, and the strips obtained heated to 140° C. The time to degrade the polymer using the indicated tin compound in this formulation was as follows:

Tin compound (with the thiodipropionate, tertiary butyl cresol and mercaptoacetate):

Time (hours)

i morcapioaccimo,	,
Dibutyl tin benzaldi (mercaptoacctate)	163
Dibutyl tin benzaldi (mercaptopropionate)	192
Dibutyl tin o-hydroxybenzaldi (mercaptoace-	
tate)	192
Dibutyl tin o-hydroxybenzaldi (mercaptopropio-	
nate)	192
Dibutyl tin butane-2,2-bis (mercaptoacetate)	192
Dibutyl tin butane-2,2-bis (mercaptopropio-	
nate)	168
	264
Dibutyl tin propanc-2,2-bis (mercaptopropio-	
nate)	192
Dibutyl tin isodecane-1.1-bis (mercaptoace-	
	168
Dibutyl tin isodecane-1,1-bis (mercaptopropio-	100
nate)	192
Dibutyl tin cyclohexane-1,1-bis (mercaptoace-	176
	264
Dibutyl tin cyclohexane-1,1-bis (mercaptopropi-	204
onate)	168
. ,	240
Dibutyl tin methylene-bis (mercaptopropio-	240
• • • • •	240
Dibutyl tin 2-ethylbutane-1,1-bis (mercapto-	240
propionate)	168
Dineopentyl tin propane-2,2-bis (mercaptopro-	100
pionale)	192
Dioctyl tin benzaldi (mercaptopropionate)	168
Dioctyl tin o-hydroxybenzaldi (mercaptoace-	100
tate)	168
Dioctyl tin o-hydroxybenzaldi (mercaptopro-	100
pionate)	120
Dioctyl tin isodecane-1,1-bis (mercaptoacetate)	168
Dioctyl tin isodecane-1,1-bis (mercaptoacetate)	100
nate)	144
Dioctyl tin propane-2,2-bis (mercaptopropio-	144
nate)	160
bis (Tributyl tin) propane-2,2-bis (mercapto-	168
	216
bis (Tributyl tin) benzaldi (mercaptopropio-	410
	212
Dibutyl tin 1-carboethoxypropane-2,2-bis (mer-	312
	72
captopropionate)	12

We claim

1. A composition of matter comprising (a) a member of the group consisting of (1) a polymer of a monoolefin having 2 to 4 carbon atoms and (2) a halogen containing resin selected from the group consisting of vinyl and vinylidene resins in which the halogen is attached directly to a carbon atom in the polymer chain and (b) a stabilizing effective amount of a member of the group consisting of (1) a hydrocarbon tin salt of a carboxy mercaptal, said salt having 1 to 3 hydrocarbon groups attached directly to the tin, said carboxy

aptal being connected to the tin atom through a xyl oxygen atom, and (2) a mixture of a member 70 the group consisting of dihydrocarbon tin oxides, dihydrocarbon tin sulfides, trihydrocarbon tin oxides, monohydrocarbon stannoic acids and monohydrocarbon tin alcoholates with a mercaptal c ntaining a free carboxyl group.

- 2. A composition f matter comprising (a) a vinyl resin in which the halogen is attached directly to a carbon atom in the polymer chain and (b) a stabilizing effective and unt of a member of the group consisting of (1) a invarocarb in the salt of a carboxy mercaptal, said salt in ring 1 to 3 hydrocarbon groups attached directly to the tin, said carb xy mercaptal being connected to the tin atom through a carboxyl oxygen atom, and (2) a mixture of a member of the group consisting of dihydrocarbon tin oxides, dihydrocarbon tin sulfides, trihydrocarbon tin oxides, monohydrocarbon stannoic acids and monohydrocarbon tin alcoholates with a mercaptal containing a free carboxyl group.
- 3. A composition according to claim 2 wherein the resin 15 is a vinyl chloride resin.
- 4. A composition of matter comprising a vinyl chloride resin and a stabilizing effective amount of a hydrocarbon in salt of a carboxy mercaptal, said salt having 1 to 3 lydrocarbon groups attached directly to the tin, said 20 carboxy mercaptal being connected to the tin atom through a carboxyl oxygen atom.

5. A composition of matter comprising a vinyl chloride resin and a stabilizing effective amount of a hydrocarbon time salt of a hydrocarbon bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic acid group.

6. A composition of matter comprising a vinyl chloride rusin and a stabilizing effective amount of a hydrocarbon tin salt of a hydroxyaryl bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic acid group.

7. A composition of matter comprising a vinyl chloride resin and a stabilizing effective amount of a mixture of (1) a hydrocarbon tin oxide and (2) a hydrocarbon bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic acid group.

3. A composition of matter comprising a vinyl chloride it in and a stabilizing effective amount of a mixture of (1) a monohydrocarbon stannoic acid and (2) a hydrocarbon bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic acid group.

40 . A composition of matter comprising a vinyl chloride resin and a stabilizing effective amount of a mixture (1) a hydrocarbon tin oxide and a monohydrocarbon stannoic acid and (2) a hydroxyaryl bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic 45 acid group.

10. A composition of matter comprising (a) a polymer of a monoclefin having 2 to 4 carbon atoms and (b) a stabilizing effective amount of a member of the group consisting of (1) a hydrocarbon tin salt of a carboxy mercaptal, said salt having 1 to 3 hydrocarbon groups attached directly to the tin, said carboxy mercaptal being connected to the tin atom through a carboxyl oxygen alom, and (2) a mixture of a member of the group consisting of dihydrocarbon tin oxides, dihydrocarbon tin oxides, trihydrocarbon tin oxides, monohydrocarbon stannoic acids and monohydrocarbon tin alcoholates with a niercaptal containing a free carboxyl group.

11. A stabilized composition according to claim 16 including a neutral sulfur compound having a thio linkage bata to a carbon atom having both a hydrogen atom and a carboxyl group attached thereto.

12. A stabilized composition according to claim 11 wherein the polymer is polypropylene and the composition also includes a phenolic antioxidant.

13. A stabilized composition according to claim 10 wherein the polymer is polypr pylene.

14. A stabilized composition according t claim 13 including a neutral sulfur compound having a thi linkage beta to a carbon atom having both a hydr gen atom and a carboxyl group attached thereto.

15. A stabilized composition according t claim 14 victorin the sulfur compound is dilauryl thiodipropionate.

15. A composition of matter comprising a solid p ly-75 propylene and a stabilizing effective amount of a hydro-

carbon tin salt of a carboxy mercaptal, said salt having 1 to 3 hydrocarbon gr ups attached directly to the tin, said carboxy mercaptal being connected t the tin atom through a carboxyl xygen atom.

17. A composition of matter c mprising a s lid poly- 5 propylene and a stabilizing effective am unt of a hydrocarbon tin salt of a hydrocarbon bis (mercaptoalkanoic acid) having 2 to 9 carbon atoms in the alkanoic acid group.

18. A composition of matter comprising a solid polypropylene and a stabilizing effective amount of a hydrocarbon tin salt of a hydroxyaryl bis (mercaptoalkanoic

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acid) having 2 t 9 carbon at ms in the alkanoic acid group.

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